



Impact of post annealing and hydrogen implantation on functional properties of Cu₂O thin films for photovoltaic applications

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ABSTRACT

Polycrystalline cuprous oxide (Cu₂O) thin films were sputtered, annealed (900 °C rapid thermal annealing) and subsequently implanted with various hydrogen ion (H⁺) doses from 5E13 to 2E15 cm⁻² with a low acceleration energy of 36 keV at room temperature to tailor the functional properties of the thin films for solar cell application. The annealed and H⁺ implanted Cu₂O thin films were post annealed at low temperatures from 100 °C to 600 °C in an inert atmosphere to promote hydrogen passivation of prevalent intrinsic acceptors and tune the carrier concentration for optimum performance as an absorption layer in a heterojunction solar cell. The H⁺ incorporation and post annealing tuned the structural, optical and electrical properties of annealed polycrystalline Cu₂O thin films. The results show an enhancement of the excitonic feature around ~2.0 eV with H⁺ dose. The normalized photoluminescence (PL) area around ~1.7 eV was drastically enhanced with increasing H⁺ doses compared to excitonic and copper vacancy related area. The normalized total PL quantum efficiency shows an enhancement in yield with elevated H⁺ doses by two orders of magnitude. The hole concentration decreases down to ~10¹³ cm⁻³, while hole mobility and resistivity increase to ~27 cm²/V and ~2.4 kΩcm, respectively, as the H⁺ implantation goes from lower to higher doses. In addition, the post annealing and H⁺ incorporation lead to a change in the energy level of the major acceptor from 0.21 eV to 0.27 eV above the valence band maximum. By following the qualitative (PL analysis) and quantitative (Hall data) outcomes, we can conclude that H⁺ implantation and post annealing likely indicates the passivation of both acceptor defects and compensating donor defects.

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1. Introduction

Hole transport semiconductor oxide materials have a great potential for sustainable energy conversion device applications (TiO₂/Ga₂O₃/Cu₂O and TiO₂/Al:ZnO/Cu₂O etc.) [1]. Cuprous oxide (Cu₂O) with ‘intrinsic’ p-type conductivity is a non-toxic, low cost, and earth-abundant solar energy material (Cu₂O/SnO₂ heterojunction for solar cell) [2]. Cu₂O has been a subject of extensive research in last few decades [3] due to its high potential as an absorber layer in a tandem hetero junction (Cu₂O/ZnO heterojunction) [4] solar cell structure based on silicon [5]. Cu₂O has also been explored for various applications, e.g. p-type thin film (TF) (ZnO/Cu₂O heterojunction) [6], p-channel Cu₂O thin-film transistors (TFTs)/complementary metal-oxide-semiconductor (CMOS) devices [7],

photovoltaic-photo electrochemical cell using p-n Cu₂O TFs and ZnO nanorods [8], polyhedral Cu₂O crystal [9] and Cu₂O TF deposited using glancing angle deposition (GLAD) technique for chemical gas sensors [10], Cu₂O-TiO₂ hybrid nanostructure for bio-sensor [11]. In addition, Cu₂O was utilized as a transparent conducting layer for monolithic perovskite CIGS tandem solar cells [12], ZnO@Au@Cu₂O nanotube arrays as efficient visible-light-driven photoelectrode [13], flexible-transparent p-n junction of N-doped Cu₂O/SnO₂ on flexible (polyethylene naphthalate) substrate [14], solar cells (ZnO/Cu₂O and/or Cu₂O/ZnO heterojunctions) [15–18]. Moreover, Cu₂O was also exploited in many other energy saving or generating semiconductor devices such as a promising cathode for thermal battery (Li/Cu₂O electrochemical cells) [19], photodetector (ITO/MoS₂/Cu₂O) with ultrahigh sensitivity [20], Cu₂O nanoparticles as a hole-injecting layer with inorganic cesium lead bromide (CsPbBr₃) perovskite light emitting diode [21], Cu₂O nanowire photocathodes for water splitting [22], and

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optoelectronic device applications [23].

However, the theoretical efficiency limit of TF based tandem hetero junction solar cell has been estimated ~30% using a top and bottom cell approach, 22% for a material of band gap 1.5 eV and 16% for a material of band gap 2.1 eV [24]. The present maximum experimental efficiency of solar cells based on p-Cu₂O:Na as an absorber combined with n-type Zn_{1-x}Ge_xO is limited to 8.1% [25] due to lack of understanding of intrinsic properties like defects [26] and doping conditions of Cu₂O [27]. Moreover, in recent years, different techniques (e.g. Cd₂SnO₄/CdS/Cu₂O/Ag using chemical technique [28], Ag:Cu₂O nanocomposites by pulsed laser [29]) and approaches (perovskite structure with Cu₂O as a hole-transporting layer for photovoltaic application [30] and modeling of CH₃NH₃Pb(I_{1-x}Cl_x)₃ based perovskite solar cell [31]) have been employed to improve the efficiency of heterojunction thin film solar cell by tuning the transport properties [32], changing the deposition or growth conditions [33], engineering the band gap (Cu₂O/Si heterojunction solar cells) [34] or nanostructure interfaces [35], sandwiching the intrinsic or intermediate layers between ZnO/Cu₂O heterojunction to annihilate the defects and enhance the performance of the device [36].

Various processes are available to reduce the defect concentration in Cu₂O TFs for instance annealing or growth in a controlled environment [37,38], by passivation using a cyanide treatment [39] or hydrogen diffusion [40]. The passivation of materials surface/bulk by hydrogen to eliminate defects and improve the electrical and optical properties is well known in the silicon solar cell and semiconductor industry [41,42]. Various techniques are also available to incorporate H⁺/H₂ into Cu₂O TFs, e.g. introduction of hydrogen during deposition [43], hydrogen plasma treatment, hydrogen ion implantation [44], and annealing of Cu₂O TFs in hydrogen ambient atmosphere. Hydrogen ions can incorporate and interact with native defects to passivate the acceptor levels through annihilation of the copper vacancy, or by forming O–H bond with the dangling bond of an O atom in the Cu₂O lattice [45]. The impact of hydrogen treatment on polycrystalline nitrogen doped Cu₂O films has previously been investigated to tune the electrical and optical properties, and the hole carrier density was controlled in the range of 10¹⁶–10¹⁸ cm⁻³ [46]. Ishizuka et al. published a comparative study on passivation of defects in polycrystalline nitrogen-doped Cu₂O films by cyanide or hydrogen treatment. The luminescence properties improved while carrier concentration varied from ~10¹⁶ – 10¹⁷ cm⁻³ [47]. Ion implantation is a well-known process in semiconductor technology and one of the most reliable techniques for hydrogen ion implantation (also called proton implantation) for passivation of surface/bulk semiconductors. However, low energy ion beam has been used to implant hydrogen ion at room temperature [48].

In the present work, the impact of post annealing temperature in Ar atmosphere and H⁺ implantation on structural, surface morphological, electrical and optical properties of annealed Cu₂O TFs have been investigated. The qualitative output acquired by photoluminescence (PL) spectroscopy has been compared with the Hall effect study. A functional correlation is established among Hall effect parameters, post annealing temperature (T_{PA}) and implanted H⁺ doses.

2. Experimental method

Cu₂O TFs were deposited by a direct current (DC) magnetron sputtering system (Semicore Triaxis) on 1 × 1 cm² quartz substrates in a controlled atmosphere (O₂/Ar ratio) at 400 °C deposition temperature. The pre-sputtering time was 10 min to avoid any contamination. The quartz substrates were cleaned in piranha solution and rinsed in deionized water, blown dry with nitrogen and

loaded into the deposition chamber. The deposition power was fixed at 100 W. The base pressure before deposition was below 5 × 10⁻⁷ Torr, while the sample stage was rotated at a constant speed of 12 rpm during deposition. The film thickness was determined using ellipsometry (J. A. Woolam alpha - SE) and modeled with the CompleteEASE software and was ~500 nm for all samples. As-grown Cu₂O films were annealed at 900 °C (in an Annealsys Micro rapid thermal annealing (RTA) furnace) for 3 min to improve the structural, optical and electrical properties. After annealing at 900 °C in RTA, the polycrystalline Cu₂O TFs were implanted with various H⁺ (Proton) of 5E13, 1E14, 4E14, 1E15 and 2E15 cm⁻² doses with an implantation energy of 36 keV to passivate the acceptor/defect levels in Cu₂O using a NEC Tandem accelerator at room temperature. The hydrogen ions implantation depth profile was simulated using SRIM code by Ziegler [49]. The X ray diffraction (XRD) pattern was recorded by a Bruker AXS D8 Discover Diffractometer, using Cu K_α-radiation and a Bragg-Brentano configuration. The UV–Vis optical transmittance spectra in the wavelength range from 400 nm to 1500 nm were measured at room temperature using a Shimadzu SolidSpec-3700 DUV spectrophotometer with an integrating sphere. The optical emission properties of TFs were investigated by photoluminescence (PL) measurements at 300 K temperature using a 488 nm Ar cw-laser with an output power of 4.4 mW as the excitation source. The focused laser beam had a power density up to ~20 kW cm⁻² in a spot of 5 μm diameter on the surface of the samples. The emission was collected by a microscope and directed via a low-pass filter (cut-off wavelength 550 nm) into an imaging spectrometer (Horiba Jobin Yvon, iHR320) coupled with an electron-multiplying CCD camera (Andor, Luca DL-658 M EMCCD) as detector. The system offered a spectral resolution of 0.2 nm. The H⁺ implanted (annealed at 900 °C in RTA) Cu₂O TFs were post annealed at 100 °C, 200 °C, 300 °C, 400 °C, 500 °C, and 600 °C in Ar atmosphere. The room temperature as well as temperature dependent Hall effect measurements (LakeShore 7604 set up) using the van-der Pauw method were carried out to determine the hole mobility, resistivity and hole carrier density. The surface morphology measurement was performed on selected samples reported in **Supplementary material (SM)** using scanning electron microscopy (JEOL JSM-IT300 SEM), equipped with a solid state back scattered electrons detector (SS-BSED) and an Everhardt–Thronley secondary electrons (SE) detector. The front surface of Cu₂O TFs was imaged at an acceleration voltage of 20 kV. The surface roughness analysis was performed on selected samples using atomic force microscopy (a Park Systems XE-70 microscope AFM), and has been discussed in detail in **SM**. The instrument is in an anechoic box to reduce background vibrations, and samples are placed onto an anti-vibration holder. The experiments were performed in non-contact mode using a non-contact high-resolution tip from Park Systems. The AFM is controlled by XEP software. The data visualization and analysis were performed using Gwyddion free and open source software [50].

3. Results and discussion

3.1. Structural, optical and microscopic analysis

The XRD pattern of as-deposited Cu₂O TFs, Cu₂O films annealed at 900 °C in RTA and H⁺ implanted Cu₂O TFs after annealing in RTA are recorded in the range 30–65° as summarized in **Fig. 1**. The XRD pattern displays that as-grown polycrystalline Cu₂O TFs is dominated by (111) reflection peak (36.65°) with other reflections peak at (200) and (220) corresponding to 42.61° and 61.34°, respectively. After annealing at 900 °C in RTA, the XRD pattern of annealed Cu₂O TFs show the preferred orientation growth in (200) reflection peak (42.61°) and higher crystallinity [51,52]. The crystallinity of the

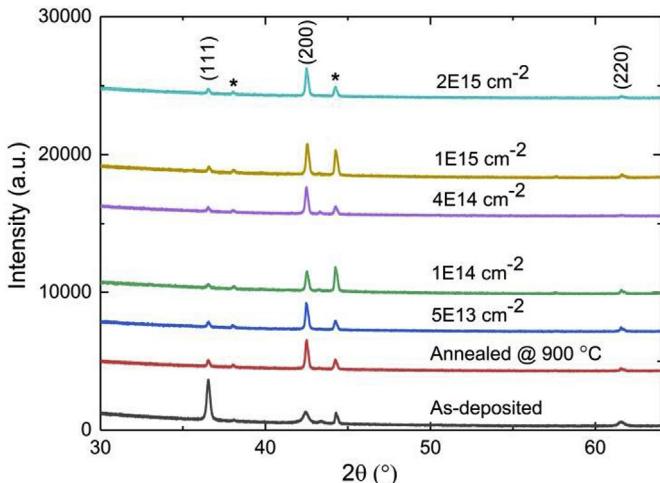


Fig. 1. XRD pattern of as-deposited Cu_2O TF (black line), Cu_2O TF annealed at 900°C in RTA (red line), and H^+ implanted Cu_2O TFs (color coded) with various H^+ doses ($5\text{E}13 - 2\text{E}15 \text{ cm}^{-2}$) after annealing at 900°C in RTA. The star (*) indicates the sample holder peak. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

films improve after annealing as the XRD peak intensity of Cu_2O increased and FWHM (Full Width Half Maximum) decreased because of larger grain sizes (SM). The phase of Cu_2O was determined by comparing the experimental XRD pattern with the Standard Powder Diffraction Cards (ICDD) patterns: 01-071-3645 or space group O_h^4 , Pn3m, No. 224 for Cu_2O [53,54]. The peak with star (*) in the XRD pattern belongs to the sample holder as displayed in Fig. 1. Hardly, any effect of H^+ implantation was recorded by XRD on the structural properties of Cu_2O TFs.

3.1.1. Optical band gap and photoluminescence (PL) spectroscopy analysis

The optical band gap of as-deposited, annealed at 900°C in RTA and H^+ implanted Cu_2O TFs were determined from the transmittance measurements (SM, Fig. S1a) using Tauc method, which assumes that absorption coefficient (α) is related to the energy band gap (E_g) by the following equation: $(\alpha h\nu) = A(h\nu - E_g)^n$, where A is a constant, $h\nu$ is photon energy and n is an index associated with specific type of optical transition (1/2 direct allowed; 3/2 direct forbidden; 2 indirect allowed; 3 indirect forbidden) [55].

Fig. 2a displays the absorption edge region of spectra plotted as $(\alpha h\nu)^2$

versus photon energy ($h\nu$), which infers a linear profile for direct band gap materials. In such Tauc plot, direct optical band gap can be determined from the intercept of the leading edge linear extrapolation with $h\nu$ axis, as displayed by dashed lines in Fig. 2a. The optical band gaps of as-deposited, annealed at 900°C and H^+ implanted Cu_2O TFs are found to be in the range from ~ 2.45 to 2.55 eV [56], which agrees well with theoretical and experimental results reported in the literature [57,58]. After annealing at 900°C in RTA, the films become more transparent and crystalline with larger grain size, all that leading to apparent blue shift of the absorption edge [59]. Next, the transmittance measurements (Fig. S1a) of H^+ implanted and post-annealed Cu_2O TFs were performed to study the impact of variable H^+ doses on the optical properties of TFs. The analysis of transmittance spectra and extracted optical band gaps suggests that Cu_2O TFs were not modified significantly within the applied H^+ dose range, which is in general agreement with the reports in literature [60,61]. This conclusion is also supported by Arita et al. report that hardly any impact on optical properties of copper and aluminum nitride TFs could be recorded after proton implantation [62].

The PL measurements (Fig. S1b) were performed at 300K on post-annealed (900°C in RTA) and post-implantation annealed (300°C) Cu_2O TFs subjected to different H^+ doses in order to investigate passivation effects of hydrogen and relaxation mechanisms of localized excitons at optically active defect centers [63]. A qualitative analysis approach has been employed to explain the observed PL spectral developments of the annealed and post-implantation annealed TFs. For this purpose, first the key emission components, excitonic and defect-related, have been designated in spectra, and then corresponding integrated PL intensities were monitored throughout different processing steps (see SM for more details). Fig. 2b summarizes PL measurements by presenting H^+ implantation dose dependencies of the integrated PL over three representative regions of spectra - I (excitonic), II (V_o^{++} and V_o^+) and III (V_{Cu}^-) defect-related emissions, respectively.

The pristine (non-implanted) material, represented by post-annealed (900°C in RTA) Cu_2O TF exhibits strong excitonic features dominating over negligible defect-related contributions in both regions II and III (Fig. S1b). Upon H^+ implantation, spectral developments in the selected three regions exhibit dissimilar trends with the increase of dose. The defect-related emission in region II (V_o^{++} and V_o^+) increases continuously with increasing H^+ dose ($2\text{E}13 - 2\text{E}15 \text{ cm}^{-2}$), suggesting that the concentration of doubly (V_o^{++}) and singly (V_o^+) ionized oxygen vacancies [64] rises up to two orders of magnitude at high doses ($2\text{E}15 \text{ cm}^{-2}$) as compared to that of pristine (non-implanted) TFs. The opposite trend is

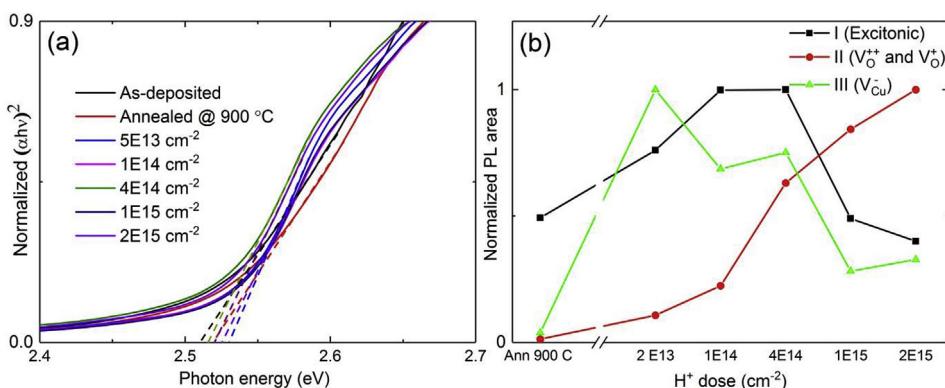


Fig. 2. (a) Optical absorption edge of as deposited (400°C), post-annealed at 900°C in RTA and H^+ implanted Cu_2O TFs. The intercepts of linear extrapolations (dashed lines) indicate direct optical band gap energies. (b) Integrated PL intensity as a function of H^+ implantation dose for three representative spectral regions: I (Excitonic), II (V_o^{++} and V_o^+) and III (V_{Cu}^-) defect-related emissions, respectively.

observed for the emission associated with copper vacancy in region III (V_{Cu}^-), which appears most prominent (ten-fold increase compared to pristine (non-implanted) TF) after the lowest-dose H^+ implantation ($2E13 \text{ cm}^{-2}$) and then decreases steadily with the increase of H^+ dose, presumably as a result of gradual suppression of copper vacancies by hydrogen. Interestingly, these inverse trends of the defect-related emissions II and III versus H^+ dose might explain the local enhancement of excitonic emission (region I) observed at around $1E14 \text{ cm}^{-2}$, which apparently represents the critical point of minimal total density of luminescent defects competing with excitons for a fixed number of the photogenerated carriers. In addition, selected Cu_2O TFs were investigated by SEM and AFM for surface morphology and roughness analysis and have been discussed in detail in the SM. No significant difference was observed on surface and morphology of selected samples after post annealing and H^+ implantation (Fig. S5 and Fig. S6).

3.2. Electrical characteristics

3.2.1. Room temperature Hall effect study

Room temperature Hall effect measurement has been performed to investigate the hole mobility, carrier density and resistivity of all Cu_2O TFs. The Hall parameters of as-deposited Cu_2O TF are $\sim 19 \text{ cm}^2/\text{V}$, $1.2 \times 10^{15} \text{ cm}^{-3}$, $268 \Omega\text{cm}$ and after annealing at 900°C in RTA, Hall effect parameters changed to $\sim 49 \text{ cm}^2/\text{V}$, $5.3 \times 10^{14} \text{ cm}^{-3}$, $627 \Omega\text{cm}$, respectively [65]. Cu_2O is natively cation-deficient and intrinsically “p-type” material due to negatively charged copper vacancies as suggested by first principles theoretical calculation [47]. The H^+ implanted (annealed at 900°C) TFs and post-implantation annealed ($100\text{--}600^\circ\text{C}$) Cu_2O TFs are also investigated by Hall measurement at room temperature as described in more detail in SM. Figs. S2 and S3 present the resistivity, hole mobility, and hole carrier concentrations for all samples. The hole carrier density and mobility have a reciprocal relationship with increasing H^+ doses. The hole mobility increases while carrier density decreases with respect to increasing H^+ doses [66]. The hole mobility is varied in the range of $\sim (1\text{--}36) \text{ cm}^2/\text{Vs}$, while hole carrier density is tuned in the range of $10^{13} \text{ -- } 10^{15} \text{ cm}^{-3}$ at different T_{PA} corresponding to various H^+ doses (Figs. S2 and S3). After H^+ incorporation films become more resistive (resistivity varied in between $\sim 0.3\text{--}5 \text{ k}\Omega\text{cm}$) as compared to as-deposited and annealed TFs [67]. Hence, we found that not only variation of H^+ doses has an impact on functional properties by controlling the hole carrier concentration (copper vacancy), but also that different T_{PA} play a leading role in influencing the electrical properties. This might be because of H^+ diffuses in or out of the Cu_2O TFs or because of trap limited conduction [68] after elevated T_{PA} as discussed in next section.

3.2.2. Impact of H^+ implantation and post-implantation annealing on transport parameters of Cu_2O TFs

Fig. 3 summarized all the results for all H^+ implanted TFs. By employing a quantitative and analytical approach on the conduction mechanism in Cu_2O TFs, a trend is established among Hall parameters when investigating the impact of various H^+ doses and different T_{PA} . Fig. 3a displays the hole carrier density as a function of H^+ doses at elevated T_{PA} . The carrier density reduces as the H^+ dose surges from lower to higher doses. The post implanted TFs exhibit the lowest carrier density at 300°C . The maximum hole carrier density is recorded $\sim 4.3 \times 10^{15} \text{ cm}^{-3}$ at 200°C T_{PA} for $5E13 \text{ cm}^{-2}$, while minimum carrier density is $\sim 9.5 \times 10^{13} \text{ cm}^{-3}$ at 300°C T_{PA} for $2E15 \text{ cm}^{-2}$. These results, where carrier density reduces with higher H^+ dose reveal that H^+ acts as a copper vacancy restrainer in p-type Cu_2O TFs, presumably which can be a possible reason for the low performance of Cu_2O based heterojunction solar

cell [69]. The drop-in carrier density with higher H^+ doses as previously discussed is in a good agreement with Tabuchi et al., where the carrier concentration was controlled by the exposure of atomic hydrogen [70]. Similar outcomes also were found by Ishizuka et al. using Si doping of Cu_2O TFs [71]. In addition, the low temperature post annealing plays a dynamic role to tune the hydrogen diffusion activity by influencing the transport parameters. The present samples are p-type, and the following condition is fulfilled $H_i^+ < V_{Cu}^-$, and most likely H^+ cations will passivate copper vacancy as mentioned in equation (Eq.) 1.



The copper vacancy formation energy is affected by cation substitution such as H^+ in this case. A substitution by a H^+ cation prompts the neutralization of copper vacancies (V_{Cu}^-) by creating the neutral V_{Cu}^0 defect and a subsequent decrease of the hole carrier density, which is likely an indication of passivation of Cu_2O TFs [72].

Fig. 3b displays a relation between T_{PA} after H^+ implantation of annealed Cu_2O TFs versus hole carrier density. The variation in hole carrier density corresponding to various H^+ doses are more visible at elevated T_{PA} (Fig. 3b). The highest hole carrier density is recorded for the lowest dose ($5E13 \text{ cm}^{-2}$ (black line)) at all T_{PA} . A drop in carrier density is recorded at 300°C for all H^+ implanted TFs as mentioned in Table 1. This is likely an indication of a favorable hydrogen activation temperature condition. The lowest carrier density is found $\sim 9.5 \times 10^{13} \text{ cm}^{-3}$ at 300°C T_{PA} for $2E15 \text{ cm}^{-2}$.

Fig. 3c and d displays a relation among hole mobility, H^+ doses and T_{PA} . No strong pattern is found in case of hole mobility versus H^+ doses (Fig. 3c). However, hole mobility improves further with higher dose, possibly due to passivation of grain boundaries or copper vacancies. A drastic change in hole mobility is recorded for all H^+ implanted Cu_2O TFs at 300°C and/or 400°C T_{PA} (Fig. 3d). The maximum hole mobility is recorded $\sim 36 \text{ cm}^2/\text{V}$ at 300°C T_{PA} for $4E14 \text{ cm}^{-2}$, while the lowest hole mobility is recorded $\sim 1 \text{ cm}^2/\text{V}$ at 200°C T_{PA} for $5E13 \text{ cm}^{-2}$. The mobility value for the highest H^+ dose $2E15 \text{ cm}^{-2}$ is $\sim 27 \text{ cm}^2/\text{V}$ at 300°C T_{PA} . In Fig. 3e and f, we investigated the impact of T_{PA} and H^+ implantation on the resistivity of Cu_2O TFs. Fig. 3e shows a variation in resistivity versus H^+ doses and Fig. 3f shows a graph between resistivity versus various T_{PA} . Unclear trend is recorded for resistivity versus H^+ doses. However, the films resistivity changed depending upon the H^+ doping and post annealing temperature. The resistivity is varied from $\sim 0.3\text{--}5 \text{ k}\Omega\text{cm}$ depending upon T_{PA} and various H^+ doses. The lowest resistivity is recorded $\sim 0.3 \text{ k}\Omega\text{cm}$ for $1E14 \text{ cm}^{-2}$ at 400°C (Fig. 3e), while the highest resistivity is obtained $\sim 5 \text{ k}\Omega\text{cm}$ for $4E14 \text{ cm}^{-2}$ at 100°C T_{PA} (Fig. 3f).

3.2.3. Temperature dependent Hall measurement (TDH)

The impact of T_{PA} on H^+ implanted and annealed Cu_2O TFs are investigated using TDH by varying the temperature from 180 to 340 K on selected samples post annealed at 300°C in Ar after H^+ implantation with $1E15$ and $2E15 \text{ cm}^{-2}$ H^+ dose, respectively. Fig. 4 displays the hole carrier density as a function of temperature. A compensated semiconductor model for a single acceptor level and a single donor level is employed to fit the quantitative behavior of charge carriers following Eq. (2) [73].

$$\frac{p(p+N_d)}{N_a - N_d - p} = \beta N_V e^{-\frac{E_a}{kT}} \quad (\text{Eq. 2})$$

where p is carrier density, N_a is acceptor concentration, N_d is compensating donor concentration, β is degeneracy level of the acceptor defect (equal to $\frac{1}{4}$), E_a is activation energy level of the defect and N_V is effective density of states of the valence band. The

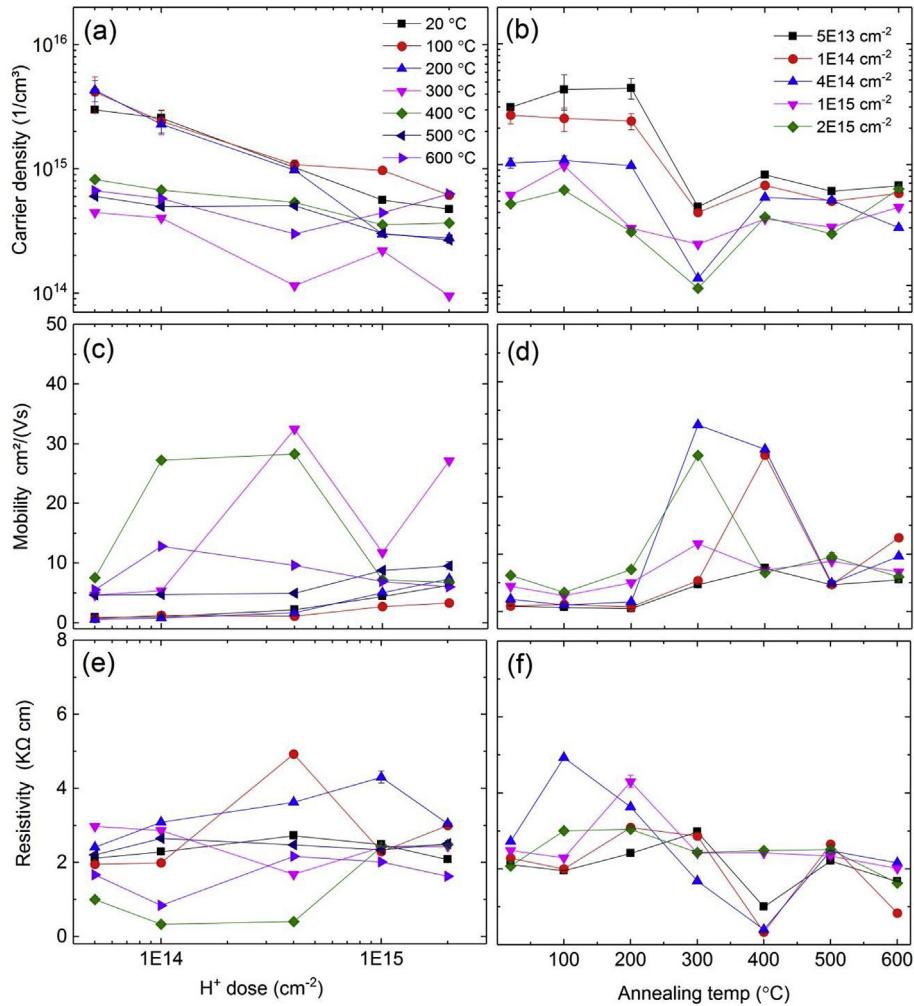


Fig. 3. (a) hole carrier density versus (vs) variation of implanted H⁺ dose for annealed, and post annealed Cu₂O TFs after H⁺ implantation; (b) hole carrier density vs variation in post annealing temperature (T_{PA}) in Ar for annealed and H⁺ implanted Cu₂O TFs; (c) hole mobility vs variation in H⁺ doses for annealed and post annealed TFs and (d) hole mobility vs post annealing temperature for annealed and H⁺ implanted Cu₂O TFs; (e) resistivity vs various H⁺ doses plot for annealed and post annealed TFs, and (f) the resistivity vs post annealing temperatures (T_{PA}) graph for annealed and H⁺ implanted Cu₂O TFs.

Table 1

Hole carrier density, mobility and resistivity values of H⁺ implanted Cu₂O TFs at 300 °C T_{PA} in Ar atmosphere.

H ⁺ doses (cm ^{-²})	5E13	1E14	4E14	1E15	2E15
Hole carrier density (cm ^{-³})	4.5×10^{14}	4.0×10^{14}	1.2×10^{14}	2.2×10^{14}	9.5×10^{13}
Hole mobility (cm ^² /Vs)	4	9	36	13	27
Resistivity (kΩcm)	3.0	2.8	1.7	2.4	2.4

quantitative fit of the experimental points to Eq. (2) (blue line in Fig. 4) revealed acceptor levels in each case at 0.21 eV (1E15 cm^{-²}) and 0.27 eV (2E15 cm^{-²}) above the VBM (SM Fig. S4) and compensation ratio (N_d/N_a) 82% (1E15 cm^{-²}) and 49% (2E15 cm^{-²}), respectively. These results are in good agreement with existing literature on the simple copper vacancies (V_{Cu}) [74–78]. A theoretical model was published by Scanlon et al. on transition levels for simple copper vacancy (0.23 eV) and split copper vacancy (0.47 eV) in p-type Cu₂O using screened hybrid-density-functional approach to investigate the formation of defect states [79]. Paul et al. studied the experimental value for two trap levels at ~0.25 eV and ~0.45 eV associated to copper vacancy using deep-level transient spectroscopy (DLTS) [70]. Kikuchi et al. also found the ionization energy of the acceptor level to be ~0.22–0.25 eV above the VBM in Cu₂O:Ni

[80]. Brochen et al. also studied temperature dependent Hall measurement on Sr doped Cu₂O films. The copper vacancy acceptor level is found at 0.27 eV above the VBM in addition to a coexisting shallow acceptor level with ionization energy about 0.13 eV, which varied with strontium doping [64]. Similarly, a change in acceptor level from 0.21 eV to 0.27 eV is found in this work when H⁺ doses varied from 1E15 to 2E15 cm^{-²}, indicating that incorporation of H⁺ leads to the acceptor levels moving towards the middle of the band gap.

Concerning the compensation ratio as mentioned in Fig. 4, a reduction in the compensation ratio can be observed by increasing the H⁺ dose from 1E15 (82%) to 2E15 (49%) cm^{-²}. The strong reduction of the compensating ratio after annealing at 300 °C with higher H⁺ dose (2E15 cm^{-²}) is probably related to a decrease in

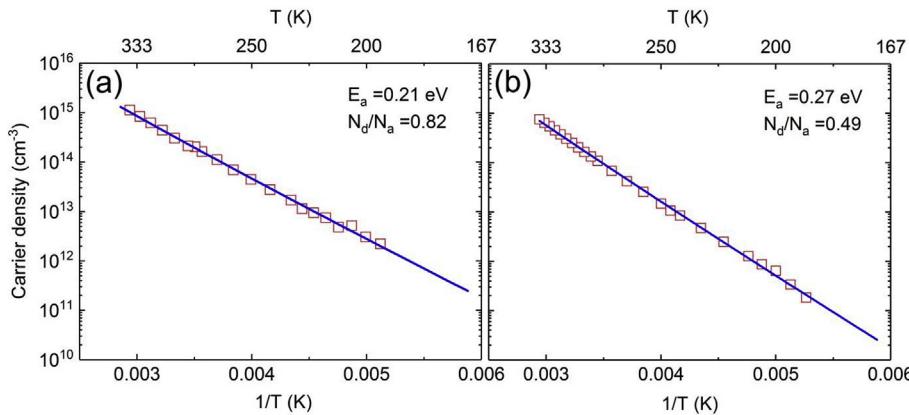


Fig. 4. Temperature dependent Hall Effect measurements of selected samples post annealed at 300 °C in Ar after H⁺ implantation, (a) the carrier density as a function of temperature for 1E15 cm⁻² H⁺ dose post annealed at 300 °C in Ar with an ionization energy level at 0.21 eV (blue line); (b) the carrier density as a function of temperature for 2E15 cm⁻² H⁺ dose post annealed at 300 °C in Ar with an ionization energy level at 0.27 eV (blue line) above the VBM. (For interpretation of the references to colour in this figure legend, the reader is referred to the Web version of this article.)

compensating defects due to a passivation in these samples, as reported by Ishizuka et al. [43]. The decrement in compensating ratio might also originate from an increase of the amount of copper vacancies, which is likely to be the most favorable intrinsic acceptors level defect in Cu₂O [47], but the reduction in carrier density with higher hydrogen ion dose (Fig. 3) suggests that the former is dominating in these samples. Hence, H⁺ implantation and subsequent annealing leads to a passivation of both acceptor defects and compensating donor defects.

4. Conclusions

The sputtered TFs of polycrystalline Cu₂O show preferential orientation growth after annealing in RTA. Post annealing and H⁺ implantation does not show any significant change on structural, optical band gap, surface and morphology of TFs, while the PL spectra and Q. E. yield were tuned subsequently with increasing H⁺ dose for all samples. However, the hole carrier density can be controlled by H⁺ implantation. The post annealed TFs (300 °C) after H⁺ implantation (2E15 cm⁻²) yielded the lowest hole carrier density of ~9.5 × 10¹³ cm⁻³. In addition, the H⁺ changes the acceptor level in the Cu₂O films from 0.21 eV to 0.27 eV when the H⁺ doping level increased from 1E15 to 2E15 cm⁻². The increment in H⁺ doses reduce the hole carrier density, indicating that H⁺ likely limits the formation of the acceptors level, probably due to the passivation behavior of the H⁺ substitution on copper vacancy site. Apparently, the analytical outcomes from PL analysis and Hall measurement culminate that post annealing and H⁺ incorporation likely indicate the passivation of both acceptor defects and compensating donor defects. Hence, H⁺ passivates both acceptors and donors in Cu₂O TFs.

Author contributions

Raj Kumar: DoE, experiment, methodology, original manuscripts writing, data analysis, Kristin Bergum: Reviewing, AFM measurement and editing manuscripts, discussion, Heine Nygard Riise: Electrical THD measurement, reviewing and editing manuscripts, discussion, Eduard Monakhov: Supervision, reviewing and editing manuscripts, discussion, project managing, Augustinas Galeckas: PL measurement, PL data analysis, reviewing and editing manuscripts, discussion.

Declaration of competing interest

There is no conflict of interest in this article.

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All authors has read and approved the manuscripts.

These is no ethical and IPR issue in this manuscripts.

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Appendix A. Supplementary data

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